9844 measured reflections

 $R_{\rm int} = 0.015$

3174 independent reflections

2534 reflections with $I > 2\sigma(I)$

Acta Crystallographica Section E **Structure Reports** Online

ISSN 1600-5368

7,9-Dimethyl-3-phenylnaphtho[1,2-b]furan-4,5-dione

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Received 4 September 2007; accepted 14 September 2007

Key indicators: single-crystal X-ray study; T = 293 K; mean σ (C–C) = 0.002 Å; R factor = 0.040; wR factor = 0.119; data-to-parameter ratio = 15.1.

In the title compound, $C_{20}H_{14}O_3$, π - π stacking interactions [3.474 (2)–3.789 (2) Å] dominate the crystal structure. The dihedral angle between the phenyl ring and fused-ring system is 32.3 (3)°.

Related literature

For related literature, see: Chang et al. (1991); Janiak (2000); Ng et al. (2000); Kongkathip et al. (2003); Park et al. (1999); Shen et al. (2005).



Experimental

Crystal data

$C_{20}H_{14}O_3$	V = 1463.0 (7) Å ³
$M_r = 302.31$	Z = 4
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation
a = 7.792 (2) Å	$\mu = 0.09 \text{ mm}^{-1}$
b = 7.602 (2) Å	T = 293 (2) K
c = 24.746 (7) Å	$0.50 \times 0.38 \times 0.27 \text{ mm}$
$\beta = 93.610 \ (5)^{\circ}$	

Data collection

Bruker SMART 1K CCD areadetector diffractometer Absorption correction: multi-scan (SADABS; Sheldrick, 1996) $T_{\rm min} = 0.956, \ T_{\rm max} = 0.976$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.040$	210 parameters
$wR(F^2) = 0.119$	H-atom parameters constrained
S = 1.03	$\Delta \rho_{\rm max} = 0.17 \ {\rm e} \ {\rm \AA}^{-3}$
3174 reflections	$\Delta \rho_{\rm min} = -0.20 \text{ e } \text{\AA}^{-3}$

Data collection: SMART (Bruker, 1999); cell refinement: SAINT-Plus (Bruker, 1999); data reduction: SAINT-Plus; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1999); software used to prepare material for publication: SHELXTL.

This work was supported by Guangdong Provincical Science Foundation.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: WW2097).

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Acta Cryst. (2007). E63, 04125 [doi:10.1107/S1600536807045059]

7,9-Dimethyl-3-phenylnaphtho[1,2-b]furan-4,5-dione

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Comment

Polycyclic *ortho*-quinonoid structures constitute a large category of significant compounds which possess a broad spectrum of biological activities such as antioxidant activity (Ng *et al.*, 2000), cytotoxicity (Park *et al.*, 1999), antifungal activity (Chang *et al.*, 1991), antineoplastic activity (Kongkathip *et al.*, 2003). Recently, we are engaged in a program directed at searching for new bioactive *ortho*-quinoid compounds. As part of our investigation (Shen *et al.*, 2005), we reported the synthesis and single-crystal structure of 7, 9-dimethyl-3-phenylnaphtho [1, 2 - b] furan-4, 5-dione (I).

The X-ray study of (I) confirms the previously proposed molecular structure based on spectroscopic data (Figure. 1).

The C—C, C=C, C—O and C=O distances show no remarkable features, except that the bond C10—C11(1.548 (2) Å) is significantly longer than normal distance of C—C bond.

A structural feature of (I) is the presence of intermolecular strong π - π stacking interactions between adjacent molecules. The geometric parameters are within the acceptable range (Janiak, 2000). The $Cg1\cdots Cg2^{i}$, $Cg1\cdots Cg3^{ii}$ and $Cg2\cdots Cg3^{ii}$ distances are 3.474 (2), 3.789 (2) and 3.690 (2)Å respectively [symmetry code: (i) -x, 1 - y, 2 - z, (ii)-x, -y, 2 - z, Cg1, Cg2 and Cg3 are centroids of the C10-containing ring, C13-containing ring and the O1-containing ring] (Figure. 2).

The naphtho[1,2-*b*]furan-4,5-dione part of the title molecule is approximately coplanar, the interplanar distances between neighbouring 13-membered ring planes are 3.458 (2) and 3.527 (2) Å.

Obviously, aromatic π - π stacking interactions play a key role in assembling the supramolecular structure.

Experimental

The title compound was prepared from 3-phenylbenzofuran-4,5-dione and N-(2-methylpenta-1,3-dienyl)acetamide by the strategy of Diels-Alder. N-(2-methylpenta-1,3-dienyl)acetamide (0.139 g, 1 mmol) and 3-phenylbenzofuran-4,5-dione (0.240 g, 1 mmol) were added to benzene (30 ml), and refluxed for 5 h. Then, silica gel (10 g, 80–120 mesh) was added to the solution and the solvent was removed in vacuum at 45°C.

The resulted dry reaction-mixture-coated silica gel was then stirred at 45°C for 24 h. Purification by chromatography on silica gel (CHCl₃/CH₃OH) gave target compound (I) in 65% yield.

Crystals of (I) suitable for X-ray analysis were grown from CDCl₃ solution.

¹H-NMR (300 MHz, CDCl₃, TMS): δ 2.35(s, 3H), 2.67(s, 3H), 7.23(s, 1H), 7.35–7.42(m, 3H), 7.59(s, 1H), 7.64 (d, 1H), 7.75(s, 1H).

¹³C NMR (75 MHz, CDCl₃): δ 21.16, 21.68, 118.71, 124.14, 127.19, 128.29, 128.30, 128.33, 128.34, 128.35, 129.09, 129.46, 129.63, 135.47, 139.33, 140.19, 140.81, 163.30, 174.84, 181.10.

ESI-MS(m/z): 303 ($[M+H]^+$).

Refinement

All H atoms were positioned geometrically and refined with a riding model, with distances 0.96 (CH₃) and 0.93Å (CH) with $U_{iso}(H) = 1.2$ (1.5 for methyl groups) times $U_{eq}(C)$.

Figures



Fig. 1. The molecular structure of (I), with atom labels and 50% probability displacement ellipsoids for non-H atoms.

Fig. 2. Part of the crystal structure of (I) formed by π - π stacking interactions. Labels *Cg* represent the centroids of rings. H atoms have been omitted for clarity. Symmetry code:(i) –*x*, 1 – *y*, 2 – *z*, (ii) –*x*, –*y*, 2 – *z*.

7,9-Dimethyl-3-phenylnaphtho[1,2-b]furan-4,5-dione

Crystal data	
$C_{20}H_{14}O_3$	$F_{000} = 632$
$M_r = 302.31$	$D_{\rm x} = 1.373 \ {\rm Mg \ m^{-3}}$
Monoclinic, $P2_1/c$	Mo K α radiation $\lambda = 0.71073$ Å
Hall symbol: -P 2ybc	Cell parameters from 1012 reflections
a = 7.792 (2) Å	$\theta = 2.8 - 26.9^{\circ}$
b = 7.602 (2) Å	$\mu = 0.09 \text{ mm}^{-1}$
c = 24.746 (7) Å	T = 293 (2) K
$\beta = 93.610 (5)^{\circ}$	Block, red
V = 1463.0 (7) Å ³	$0.50 \times 0.38 \times 0.27 \text{ mm}$
Z = 4	

Data collection

Bruker SMART 1K CCD area-detector diffractometer	3174 independent reflections
Radiation source: fine-focus sealed tube	2534 reflections with $I > 2\sigma(I)$
Monochromator: graphite	$R_{\rm int} = 0.015$
T = 293(2) K	$\theta_{\text{max}} = 27.1^{\circ}$

φ and ω scans	$\theta_{\min} = 2.6^{\circ}$
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	$h = -9 \rightarrow 9$
$T_{\min} = 0.956, T_{\max} = 0.976$	$k = -9 \rightarrow 9$
9844 measured reflections	$l = -31 \rightarrow 30$

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.040$	H-atom parameters constrained
$wR(F^2) = 0.119$	$w = 1/[\sigma^2(F_o^2) + (0.0585P)^2 + 0.3061P]$ where $P = (F_o^2 + 2F_c^2)/3$
<i>S</i> = 1.03	$(\Delta/\sigma)_{\rm max} = 0.001$
3174 reflections	$\Delta \rho_{max} = 0.17 \text{ e } \text{\AA}^{-3}$
210 parameters	$\Delta \rho_{min} = -0.20 \text{ e } \text{\AA}^{-3}$
Primary atom site location: structure-invariant direct	

Primary atom site location: structure-invariant direct methods Extinction correction: none

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted *R*-factor *wR* and goodness of fit S are based on F^2 , conventional *R*-factors *R* are based on F, with F set to zero for negative F^2 . The threshold expression of $F^2 > 2 \operatorname{sigma}(F^2)$ is used only for calculating *R*-factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2 are statistically about twice as large as those based on F, and R– factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (A^2)

	x	У	Ζ	$U_{iso}*/U_{eq}$
C1	0.33460 (19)	0.0790 (2)	0.94412 (6)	0.0528 (4)
H1	0.4424	0.0311	0.9392	0.063*
C2	0.21428 (17)	0.10566 (18)	0.90371 (5)	0.0454 (3)
C3	0.24088 (19)	0.06386 (18)	0.84666 (5)	0.0482 (3)
C4	0.4054 (2)	0.0779 (2)	0.82849 (6)	0.0577 (4)
H4	0.4939	0.1223	0.8515	0.069*
C5	0.4389 (2)	0.0267 (3)	0.77658 (7)	0.0717 (5)
H5	0.5498	0.0361	0.7650	0.086*
C6	0.3096 (3)	-0.0379 (3)	0.74210 (7)	0.0749 (5)
H6	0.3328	-0.0736	0.7074	0.090*
C7	0.1453 (3)	-0.0498 (2)	0.75908 (7)	0.0695 (5)
H7	0.0574	-0.0931	0.7356	0.083*
C8	0.1096 (2)	0.0023 (2)	0.81090 (6)	0.0579 (4)

H8	-0.0023	-0.0040	0.8218	0.069*
C9	0.06983 (16)	0.17866 (17)	0.93019 (5)	0.0422 (3)
C10	-0.09938 (18)	0.23874 (19)	0.91119 (6)	0.0486 (3)
C11	-0.21411 (17)	0.30834 (18)	0.95534 (6)	0.0493 (3)
C12	-0.14654 (17)	0.31760 (17)	1.01237 (5)	0.0440 (3)
C13	-0.25251 (18)	0.38256 (18)	1.05073 (6)	0.0500 (3)
H13	-0.3638	0.4180	1.0401	0.060*
C14	-0.19467 (19)	0.39517 (18)	1.10448 (6)	0.0505 (3)
C15	-0.02604 (19)	0.34631 (18)	1.11827 (6)	0.0496 (3)
H15	0.0150	0.3590	1.1542	0.059*
C16	0.08540 (17)	0.27926 (17)	1.08151 (5)	0.0450 (3)
C17	0.02245 (16)	0.26205 (16)	1.02753 (5)	0.0405 (3)
C18	0.11923 (16)	0.19215 (17)	0.98400 (5)	0.0412 (3)
C19	-0.3093 (2)	0.4566 (2)	1.14715 (7)	0.0664 (4)
H19A	-0.4067	0.5171	1.1303	0.100*
H19B	-0.2465	0.5349	1.1715	0.100*
H19C	-0.3480	0.3570	1.1669	0.100*
C20	0.2641 (2)	0.2282 (2)	1.10205 (6)	0.0600 (4)
H20A	0.2802	0.2562	1.1399	0.090*
H20B	0.3465	0.2916	1.0824	0.090*
H20C	0.2799	0.1041	1.0971	0.090*
O1	0.28067 (12)	0.13026 (14)	0.99344 (4)	0.0518 (3)
O2	-0.15827 (15)	0.23743 (19)	0.86475 (4)	0.0759 (4)
O3	-0.35925 (14)	0.35313 (18)	0.94093 (5)	0.0736 (4)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0484 (7)	0.0687 (9)	0.0413 (7)	0.0050 (7)	0.0042 (6)	-0.0011 (7)
C2	0.0484 (7)	0.0455 (7)	0.0423 (7)	-0.0051 (6)	0.0027 (5)	0.0041 (6)
C3	0.0602 (8)	0.0444 (7)	0.0398 (7)	-0.0017 (6)	0.0017 (6)	0.0050 (6)
C4	0.0627 (9)	0.0629 (9)	0.0479 (8)	-0.0017 (7)	0.0065 (7)	-0.0002 (7)
C5	0.0794 (12)	0.0828 (12)	0.0548 (10)	0.0027 (9)	0.0181 (8)	0.0006 (9)
C6	0.1036 (14)	0.0796 (12)	0.0421 (9)	0.0043 (10)	0.0096 (9)	-0.0019 (8)
C7	0.0918 (13)	0.0705 (11)	0.0444 (8)	-0.0075 (9)	-0.0095 (8)	0.0001 (8)
C8	0.0670 (9)	0.0598 (9)	0.0460 (8)	-0.0077 (7)	-0.0029(7)	0.0057 (7)
C9	0.0450 (7)	0.0410 (7)	0.0401 (7)	-0.0064 (5)	-0.0009 (5)	0.0063 (5)
C10	0.0491 (7)	0.0497 (7)	0.0459 (8)	-0.0045 (6)	-0.0056 (6)	0.0082 (6)
C11	0.0431 (7)	0.0486 (8)	0.0555 (8)	-0.0039 (6)	-0.0036 (6)	0.0087 (6)
C12	0.0439 (7)	0.0393 (7)	0.0486 (8)	-0.0061 (5)	0.0022 (5)	0.0063 (5)
C13	0.0443 (7)	0.0456 (7)	0.0603 (9)	-0.0009 (6)	0.0055 (6)	0.0046 (6)
C14	0.0554 (8)	0.0393 (7)	0.0577 (9)	-0.0035 (6)	0.0113 (6)	0.0000 (6)
C15	0.0579 (8)	0.0457 (7)	0.0452 (7)	-0.0057 (6)	0.0039 (6)	-0.0020 (6)
C16	0.0487 (7)	0.0418 (7)	0.0441 (7)	-0.0058 (5)	0.0003 (6)	0.0016 (5)
C17	0.0424 (7)	0.0366 (6)	0.0425 (7)	-0.0063 (5)	0.0023 (5)	0.0050 (5)
C18	0.0403 (6)	0.0414 (7)	0.0415 (7)	-0.0044 (5)	-0.0001 (5)	0.0056 (5)
C19	0.0674 (10)	0.0653 (10)	0.0681 (11)	0.0055 (8)	0.0161 (8)	-0.0080 (8)
C20	0.0544 (8)	0.0803 (11)	0.0442 (8)	0.0030 (8)	-0.0051 (6)	-0.0052 (7)

01	0.0449 (5)	0.0695 (6)	0.0404 (5)	0.0060 (4)	-0.0017 (4)	-0.0002 (4)
02	0.0706 (7)	0.1068 (10)	0.0479 (6)	0.0163 (7)	-0.0150 (5)	0.0029 (6)
O3	0.0486 (6)	0.1003 (10)	0.0703 (8)	0.0121 (6)	-0.0092 (5)	0.0031 (7)
Geometric pa	rameters (Å, °)					
C1—C2		1.342 (2)	C11-	03	1.2	130 (17)
C101		1.3721 (17)	C11-	C12	1.47	76 (2)
C1—H1		0.9300	C12-	C13	1.38	88 (2)
С2—С9		1.4480 (19)	C12-	C17	1.4	110 (19)
C2—C3		1.474 (2)	C13-	C14	1.38	81 (2)
C3—C4		1.389 (2)	C13-	-H13	0.93	300
C3—C8		1.391 (2)	C14-	C15	1.38	87 (2)
C4—C5		1.383 (2)	C14-	C19	1.50	00 (2)
C4—H4		0.9300	C15-	C16	1.39	93 (2)
C5—C6		1.370 (3)	C15-	-H15	0.9.	300
С5—Н5		0.9300	C16-	C17	1.39	999 (19)
C6—C7		1.375 (3)	C16-	C20	1.50	03 (2)
С6—Н6		0.9300	C17-	C18	1.45	538 (18)
С7—С8		1.387 (2)	C18-	01	1.34	496 (16)
С7—Н7		0.9300	C19-	—Н19А	0.90	500
С8—Н8		0.9300	C19-	-H19B	0.90	500
C9—C18		1.3666 (18)	C19-	-H19C	0.90	500
C9—C10		1.4457 (19)	C20-	H20A	0.90	500
C10—O2		1.2105 (17)	C20-	-H20B	0.90	500
C10—C11		1.548 (2)	C20-	-H20C	0.90	500
C2—C1—O1		112.39 (13)	C13-		118	.62 (13)
C2—C1—H1		123.8	C17-		120	.70 (12)
01—C1—H1		123.8	C14-	C13C12	120	.77 (13)
C1—C2—C9		104.32 (12)	C14-	—С13—Н13	119	.6
C1—C2—C3		123.16 (13)	C12-		119	.6
C9—C2—C3		132.53 (12)	C13-		117	.72 (13)
C4—C3—C8		118.45 (14)	C13-		121	.81 (14)
C4—C3—C2		118.80 (13)	C15-		120	.46 (14)
C8-C3-C2		122.66 (14)	C14-	-C15-C16	123	.84 (14)
$C_{5} - C_{4} - C_{3}$		120.75 (16)	C14-	-C15-H15	118	.1
C3—C4—H4		119.6	C16-	-C15H15	118	.1
C3-C4-H4		119.0	C15-	-C16-C17	11/	.52 (13)
C6 - C5 - U5		120.33 (17)	C13-	-C16-C20	118	.41(13)
C0-C3-H5		119.0	C1/-	-C10-C20	124	$\frac{10}{12}$
$C_{4} C_{5} C_{6} C_{7}$		119.8	C16-	-C17-C18	119	.40(12)
С5—С6—Н6		119.72 (17)	C10-	-C17-C18	125	55 (12)
С3—С6—Н6		120.1	01_	-C18-C9	115	17(11)
$C_{1} = C_{0} = C_{10}$		120.1	01-	-C18C17	110	37 (11)
С6—С7—Н7		119 7	C9	-C18C17	121	46 (12)
С8—С7—Н7		119.7	C14-	-C19-H19A	109	5
C7-C8-C3		120 15 (16)	C14-	-C19-H19R	109	
С7—С8—Н8		119.9	H194	4—C19—H19B	109	.5
		/ / /			10)	

С3—С8—Н8	119.9	C14—C19—H19C	109.5
C18—C9—C10	119.23 (12)	H19A—C19—H19C	109.5
C18—C9—C2	106.88 (11)	H19B—C19—H19C	109.5
C10—C9—C2	133.87 (12)	C16—C20—H20A	109.5
O2—C10—C9	126.13 (14)	С16—С20—Н20В	109.5
O2-C10-C11	118.14 (13)	H20A—C20—H20B	109.5
C9—C10—C11	115.72 (12)	C16—C20—H20C	109.5
O3—C11—C12	122.51 (14)	H20A—C20—H20C	109.5
O3—C11—C10	117.24 (13)	H20B-C20-H20C	109.5
C12-C11-C10	120.25 (12)	C18—O1—C1	106.23 (10)
C13—C12—C17	120.68 (13)		
O1—C1—C2—C9	-0.80 (17)	C10-C11-C12-C17	-0.52 (19)
O1—C1—C2—C3	179.16 (12)	C17—C12—C13—C14	0.4 (2)
C1—C2—C3—C4	-31.4 (2)	C11—C12—C13—C14	-179.45 (12)
C9—C2—C3—C4	148.53 (15)	C12—C13—C14—C15	2.1 (2)
C1—C2—C3—C8	145.23 (16)	C12—C13—C14—C19	-176.95 (13)
C9—C2—C3—C8	-34.8 (2)	C13-C14-C15-C16	-2.6 (2)
C8—C3—C4—C5	-2.0 (2)	C19-C14-C15-C16	176.49 (14)
C2—C3—C4—C5	174.80 (15)	C14-C15-C16-C17	0.5 (2)
C3—C4—C5—C6	0.4 (3)	C14-C15-C16-C20	-178.85 (13)
C4—C5—C6—C7	0.8 (3)	C15-C16-C17-C12	2.02 (18)
C5—C6—C7—C8	-0.3 (3)	C20-C16-C17-C12	-178.68 (13)
C6—C7—C8—C3	-1.3 (3)	C15-C16-C17-C18	-178.94 (12)
C4—C3—C8—C7	2.4 (2)	C20-C16-C17-C18	0.4 (2)
C2—C3—C8—C7	-174.24 (14)	C13—C12—C17—C16	-2.48 (19)
C1—C2—C9—C18	1.27 (15)	C11—C12—C17—C16	177.34 (12)
C3—C2—C9—C18	-178.68 (14)	C13—C12—C17—C18	178.39 (11)
C1—C2—C9—C10	179.28 (15)	C11—C12—C17—C18	-1.80 (17)
C3—C2—C9—C10	-0.7 (3)	C10-C9-C18-O1	-179.68 (11)
C18—C9—C10—O2	178.62 (15)	C2-C9-C18-O1	-1.32 (14)
C2—C9—C10—O2	0.8 (3)	C10-C9-C18-C17	-0.2 (2)
C18—C9—C10—C11	-2.22 (18)	C2-C9-C18-C17	178.16 (12)
C2-C9-C10-C11	179.96 (13)	C16-C17-C18-O1	2.71 (19)
O2-C10-C11-O3	2.1 (2)	C12-C17-C18-O1	-178.22 (11)
C9—C10—C11—O3	-177.14 (13)	C16—C17—C18—C9	-176.72 (13)
O2-C10-C11-C12	-178.18 (14)	C12—C17—C18—C9	2.35 (19)
C9—C10—C11—C12	2.59 (19)	C9—C18—O1—C1	0.84 (15)
O3—C11—C12—C13	-1.0 (2)	C17—C18—O1—C1	-178.69 (12)
C10-C11-C12-C13	179.29 (12)	C2-C1-O1-C18	0.02 (17)
O3—C11—C12—C17	179.19 (14)		



Fig. 1



